

O₂(¹Δ) PRODUCTION IN HIGH PRESSURE FLOWING He/O₂ PLASMAS: SCALING AND QUENCHING

Natalia Y. Babaeva, Ramesh Arakoni, and Mark J. Kushner
Iowa State University
Ames, IA 50011, USA

natalie5@iastate.edu arakoni@iastate.edu mjk@iastate.edu
<http://uigelz.ece.iastate.edu>

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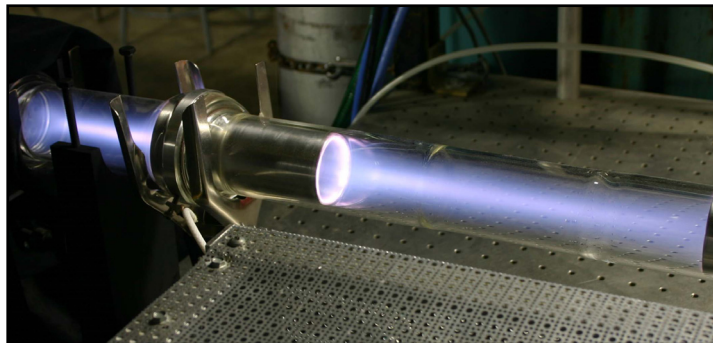
*** Work supported by Air Force Office of Scientific Research and National Science Foundation.**

AGENDA

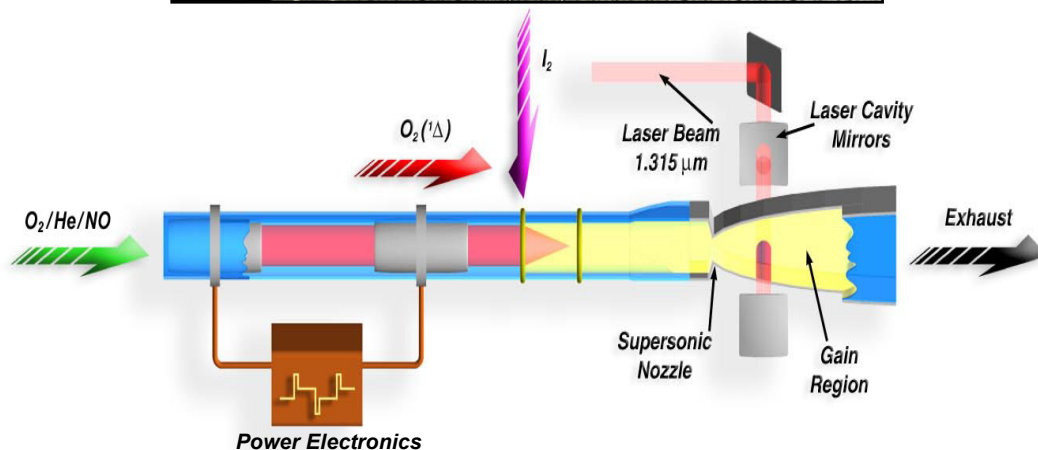
- Introduction to eCOIL
- Description of the models (GlobalKIN, 2-d nonPDPSIM)
- Performance at high pressures
 - Sublinear scaling of $O_2(^1\Delta)$ with pressure
 - Main quenchers of $O_2(^1\Delta)$
 - Gas heating and discharge stability
 - CW vs Spiker-Sustainer excitation
- Concluding Remarks

ELECTRICALLY EXCITED OXYGEN-IODINE LASERS

- In chemical oxygen-iodine lasers (COILs), oscillation at $1.315\ \mu\text{m}$ $I(^2P_{1/2}) \rightarrow I(^2P_{3/2})$ occurs by excitation transfer of $\text{O}_2(^1\Delta)$ to I_2 and I .
- Plasma production of $\text{O}_2(^1\Delta)$ in electrical COILs (eCOILs) eliminates liquid phase generators.



- I_2 injection and supersonic expansion (to lower T_{gas} for inversion) occurs downstream of the plasma zone.



- Ref: CU Aerospace

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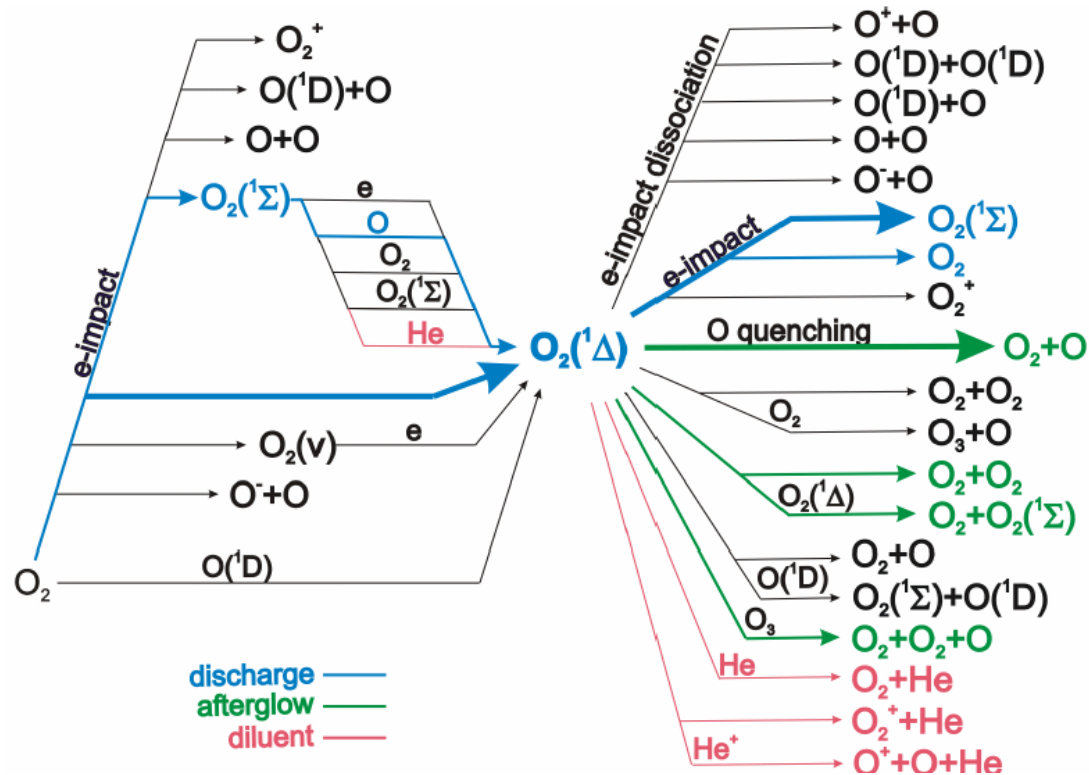
HIGH PRESSURE ISSUES FOR $O_2(^1\Delta)$ PRODUCTION

- Motivation for operating eCOIL at higher pressures
 - Larger absolute density of $O_2(^1\Delta)$ for a given yield (fraction of O_2 in excited state).
 - Higher back pressure for expansion.
- If there are no second order effects....
 - If Power \sim pressure and flowrate \sim pressure then eV/molecule is a constant.
 - $O_2(^1\Delta)$ production scales linearly with pressure.
 - Yield remains constant.
- Electron temperature and gas temperature weakly depend upon pressure.

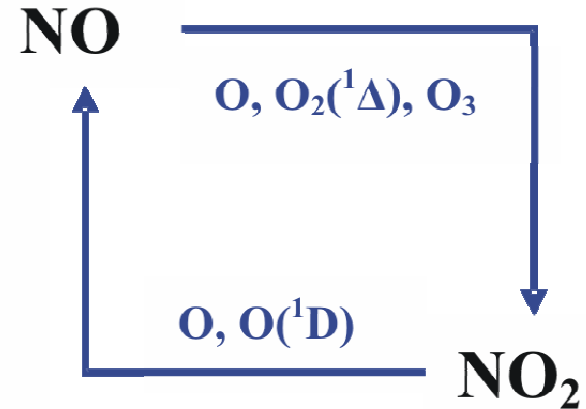
SECOND ORDER EFFECTS

- Second order effects may dominate the scaling of $O_2(^1\Delta)$ at higher pressures
- Three-body reactions produce quenchers of $O_2(^1\Delta)$, primarily O_3
- Quenching rates by O may increase with pressure with 3-body enhancement.
- Reactions $A + B + M \rightarrow AB + M$ are usually exothermic - significant heating source.
- Maintaining stability of discharge becomes more difficult at higher pressures.
- In this talk, we report on computational investigation of importance of second order effects in pressure scaling.

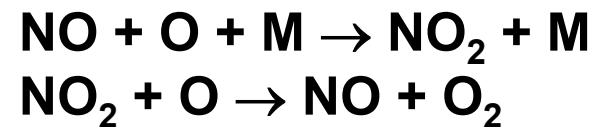
O₂(¹Δ) KINETICS IN He/O₂ DISCHARGES



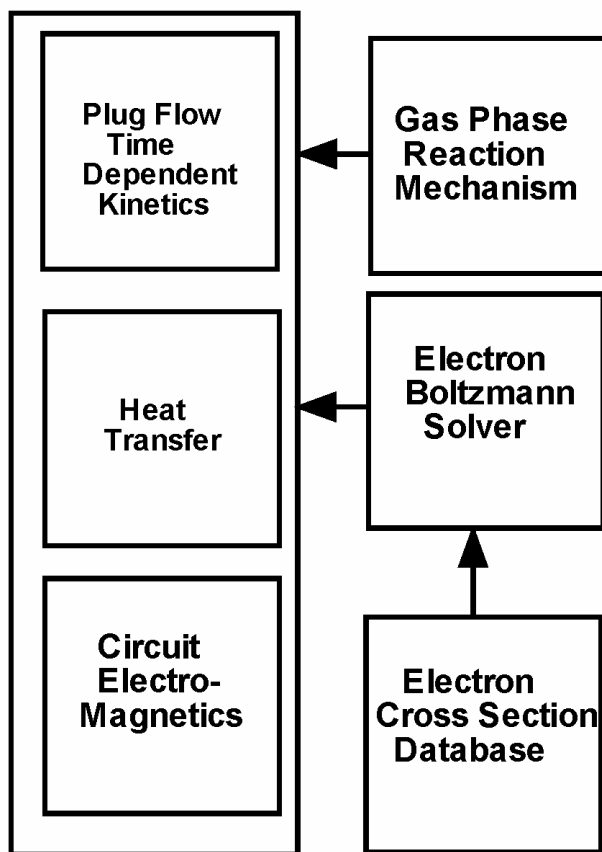
- Direct electron impact of O₂ and excitation of O₂(¹Σ) (quenching) are main channels of O₂(¹Δ) production.
- O and O₃ quench O₂(¹Δ).



- NO/NO₂ recycling chain scavenges O atoms:



DESCRIPTION OF GlobalKIN



- **Global model utilized in a plug flow mode.**
- **Boltzmann's equation solved for electron energy distribution.**
- **Ion transport linked to database.**
- **Electric field obtained from circuit model or electro-magnetics power balance.**
- **Plug flow model includes enthalpy induced change in flow speeds.**

DESCRIPTION OF 2d-nonPDPSIM: CHARGED PARTICLES, SOURCES

- Poisson's equation, continuity equations and surface charge are simultaneously solved using a Newton iteration technique.

$$-\nabla \cdot \epsilon \nabla \Phi = \sum_j N_j q_j + \rho_s$$

$$\frac{\partial N_j}{\partial t} = -\nabla \cdot \vec{\phi}_j + S_j$$

$$\frac{\partial \rho_s}{\partial t} = \sum_j -q_j (\nabla \cdot \vec{\phi}_j + S_j) - \nabla \cdot (\sigma(-\nabla \Phi))$$

- Electron energy equation:

$$\frac{\partial (n_e \epsilon)}{\partial t} = \vec{j} \cdot \vec{E} - n_e \sum_i \Delta \epsilon_i N_i \kappa_i - \nabla \cdot \left(\frac{5}{2} \epsilon \nabla \Phi - \lambda \nabla T_e \right), \quad \vec{j} = q \vec{\phi}_e$$

DESCRIPTION OF 2d-nonPDPSIM: NEUTRAL PARTICLE TRANSPORT

- Fluid averaged values of mass density, mass momentum and thermal energy density obtained using unsteady algorithms.

$$\frac{\partial \rho}{\partial t} = -\nabla \cdot (\rho \vec{v}) + (\text{inlets, pumps})$$

$$\frac{\partial(\rho \vec{v})}{\partial t} = \nabla(NkT) - \nabla \cdot (\rho \vec{v} \vec{v}) - \nabla \cdot \bar{\mu} + \sum_i q_i N_i \vec{E}_i$$

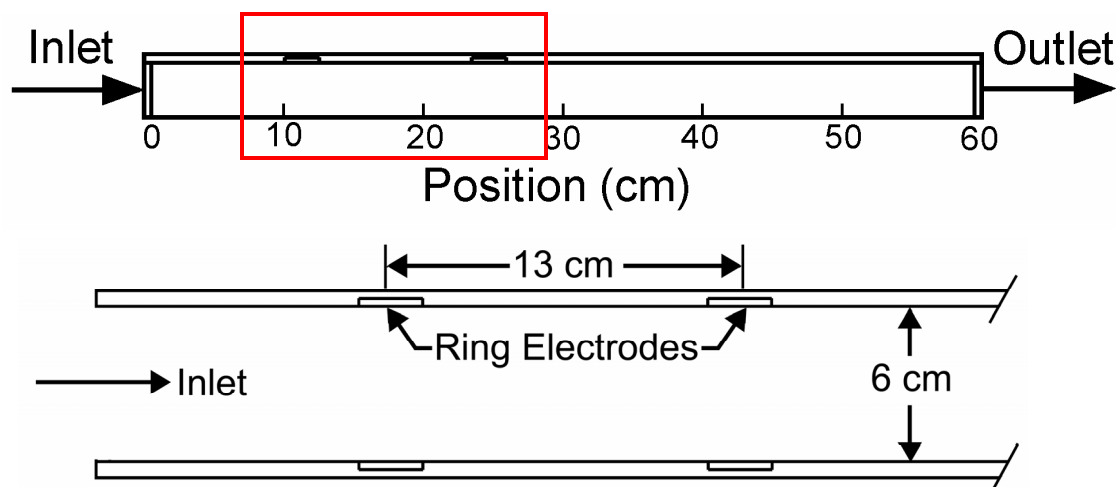
$$\frac{\partial(\rho c_p T)}{\partial t} = -\nabla(-\kappa \nabla T + \rho \vec{v} c_p T) + P_i \nabla \cdot \mathbf{v}_f - \sum_i R_i \Delta H_i + \sum_i \vec{j}_i \cdot \vec{E}$$

- Individual fluid species diffuse in the bulk fluid.

$$N_i(t + \Delta t) = N_i(t) - \nabla \cdot \left(\vec{v}_f - D_i N_T \nabla \left(\frac{N_i(t + \Delta t)}{N_T} \right) \right) + S_V + S_S$$

GEOMETRY FOR CAPACITIVE EXCITATION

He/O₂/NO



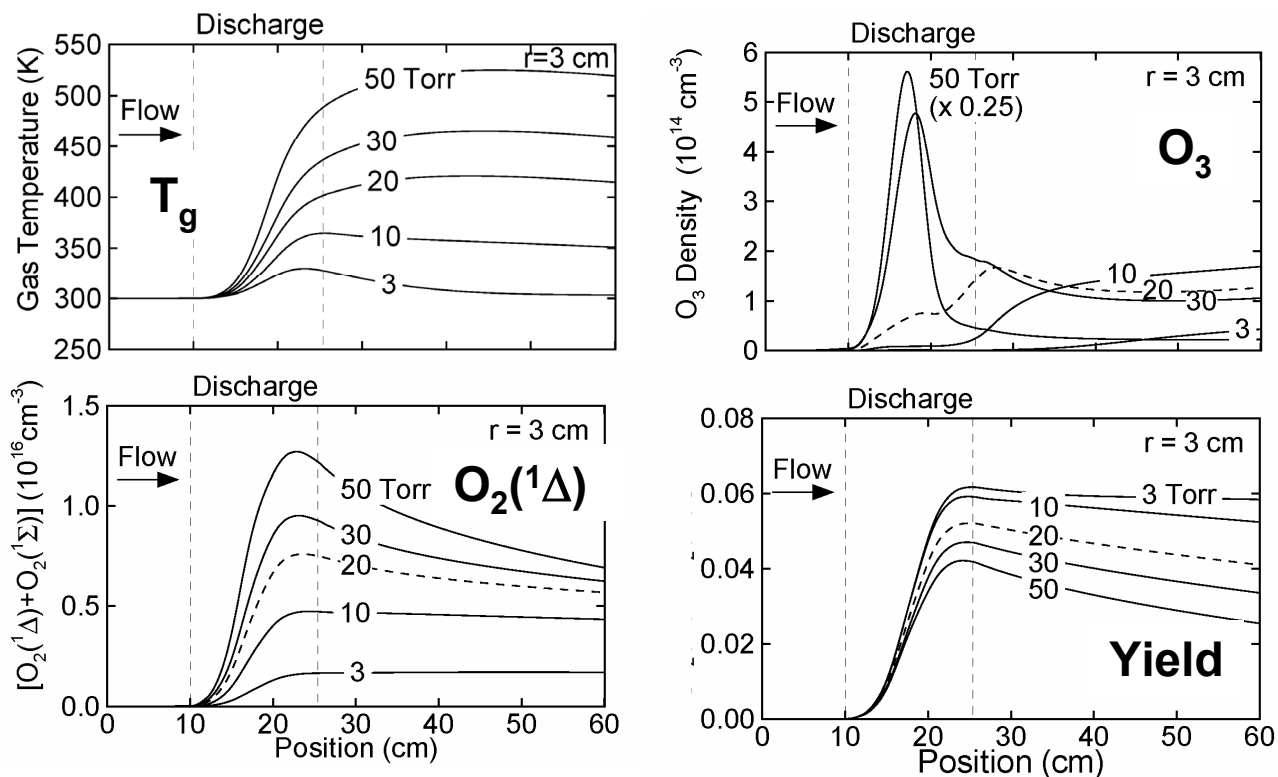
- **Yield**

$$Y = \frac{[O_2(^1\Delta)] + [O_2(^1\Sigma)]}{([O_2] + [O_2(v)] + [O_2(^1\Delta)] + [O_2(^1\Sigma)] + 0.5[O] + 1.5[O_3])}$$

- **Cylindrical tube (2 and 6 cm diameter)**
- **Capacitive excitation (10-25 MHz).**
- **Constant eV/molecule and residence time.**
- **He/O₂=70/30, 3-50 Torr, 10's to 2000 W**
- **NO injection to control O atom inventory.**

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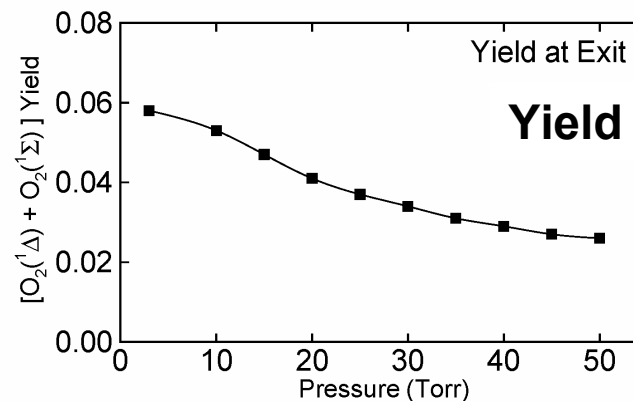
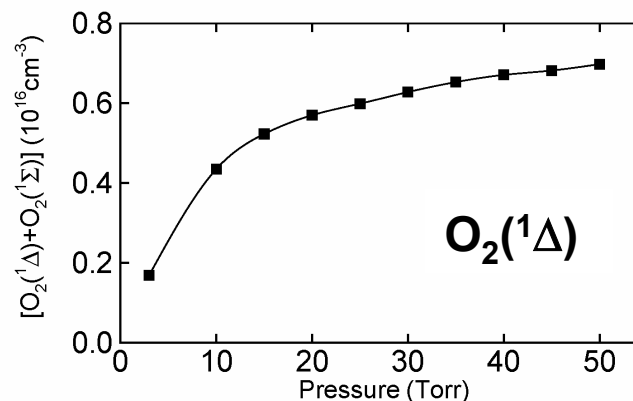
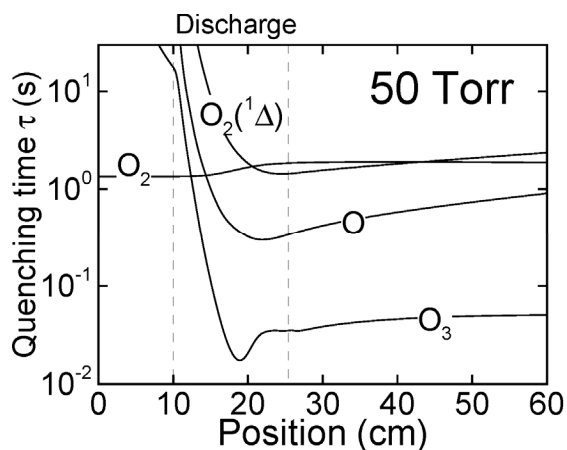
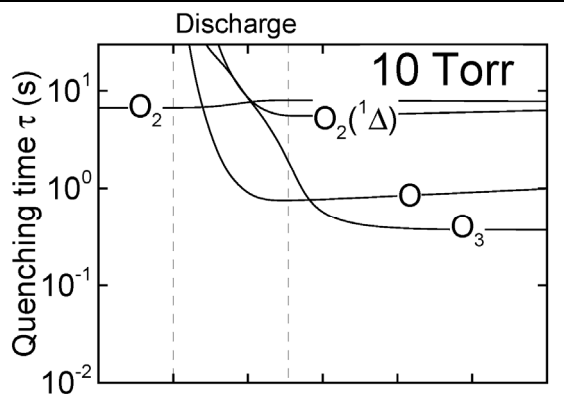
2nd ORDER EFFECT: 3-BODY REACTIONS AND O₃



- Gas heating due to exothermic 3-body reactions.
- Peak O₃ density downstream at low pressure; in discharge region at higher pressures due to O₃ destruction with increase of T_g
- Sub-linear scaling of O₂(¹Δ) yield due to quenching by O₃.
 - GlobalKin: 3-50 Torr, 40-670 W, 0.3 eV/molecule

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MAIN QUENCHERS OF $O_2(^1\Delta)$: OZONE



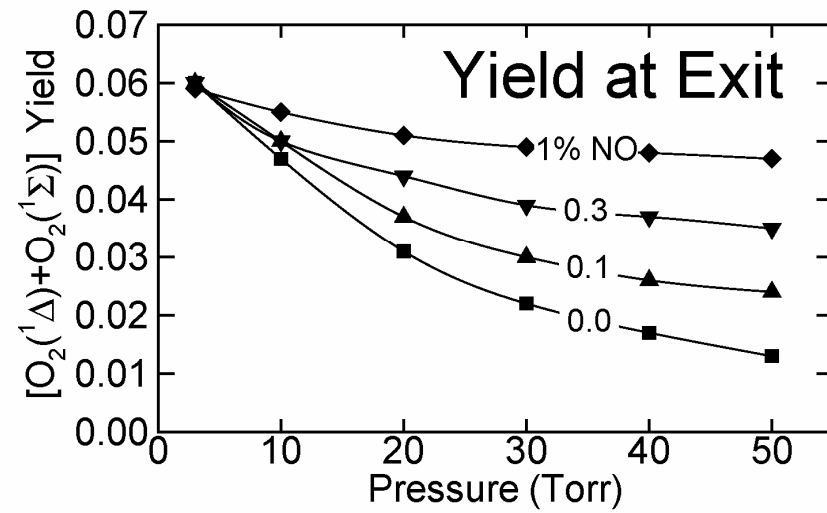
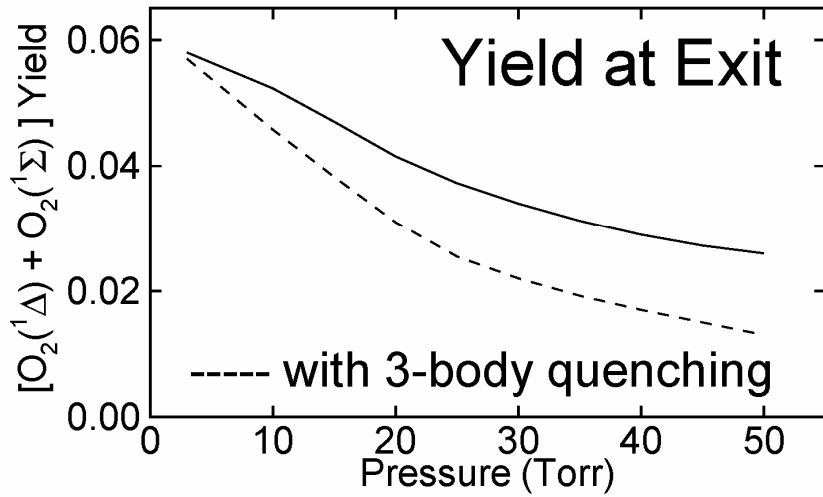
- O_3 is the primary quencher > 10 Torr

- Sublinear $O_2(^1\Delta)$ scaling and decrease of yield with pressure.

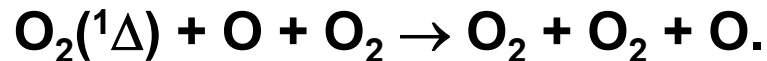
- GlobalKin: 3-50 Torr, 40-670 W, 0.3 eV/molecule

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MAIN QUENCHERS OF $O_2(^1\Delta)$: ATOMIC OXYGEN



- Yields decrease by a factor of 2 at 50 Torr when including three-body quenching:



A. N. Vasilieva et. al.
JPD 37, 2455 (2004)

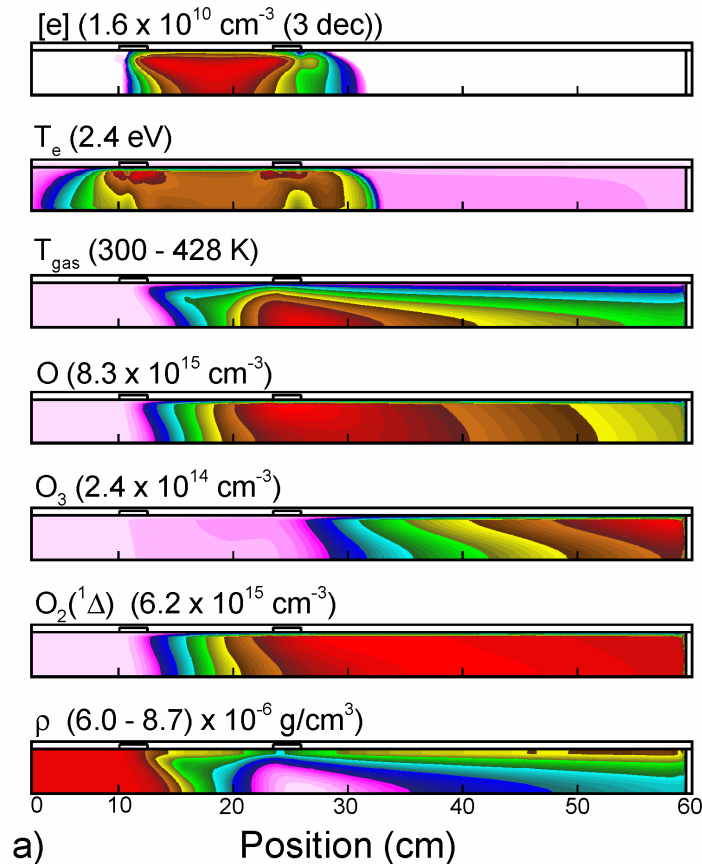
- Addition of small amounts (< 1%) of NO in inlet scavenges O atoms and restores yield.

- GlobalKin: 3-50 Torr, 40-670 W, 0.3 eV/molecule

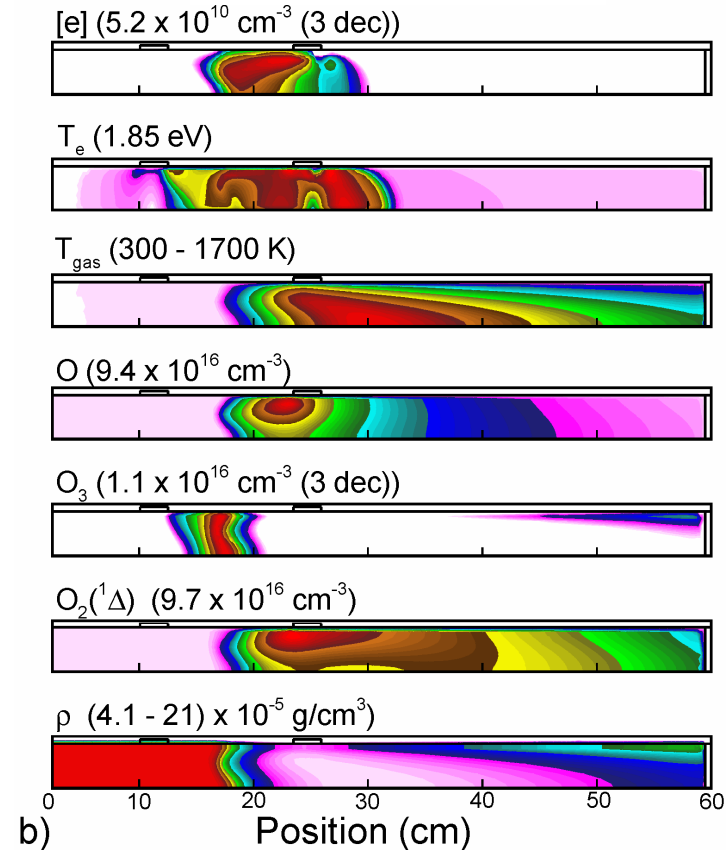
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PLASMA PARAMETERS 10 - 50 TORR

• 10 Torr, 133 W



• 50 Torr, 670 W

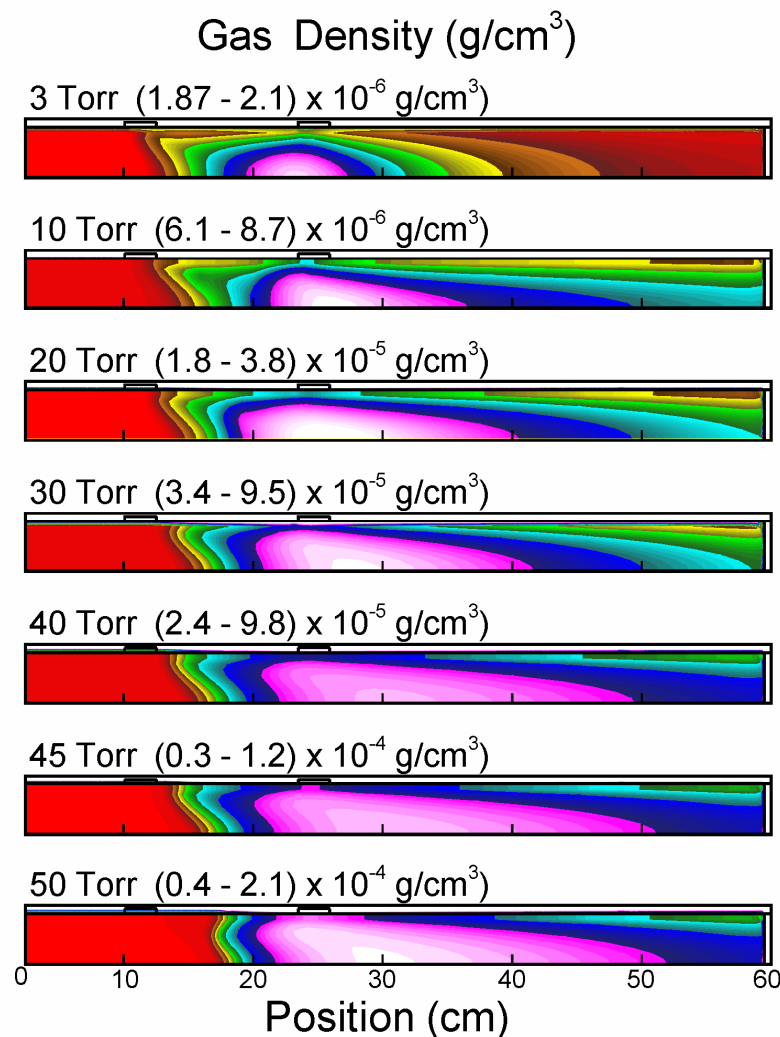


- **Gas heating increases with pressure due to 3-body reactions resulting in non-uniform power deposition.**

MIN  MAX

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RAREFACTION DUE TO 3-BODY ENTHALPY CHANGES

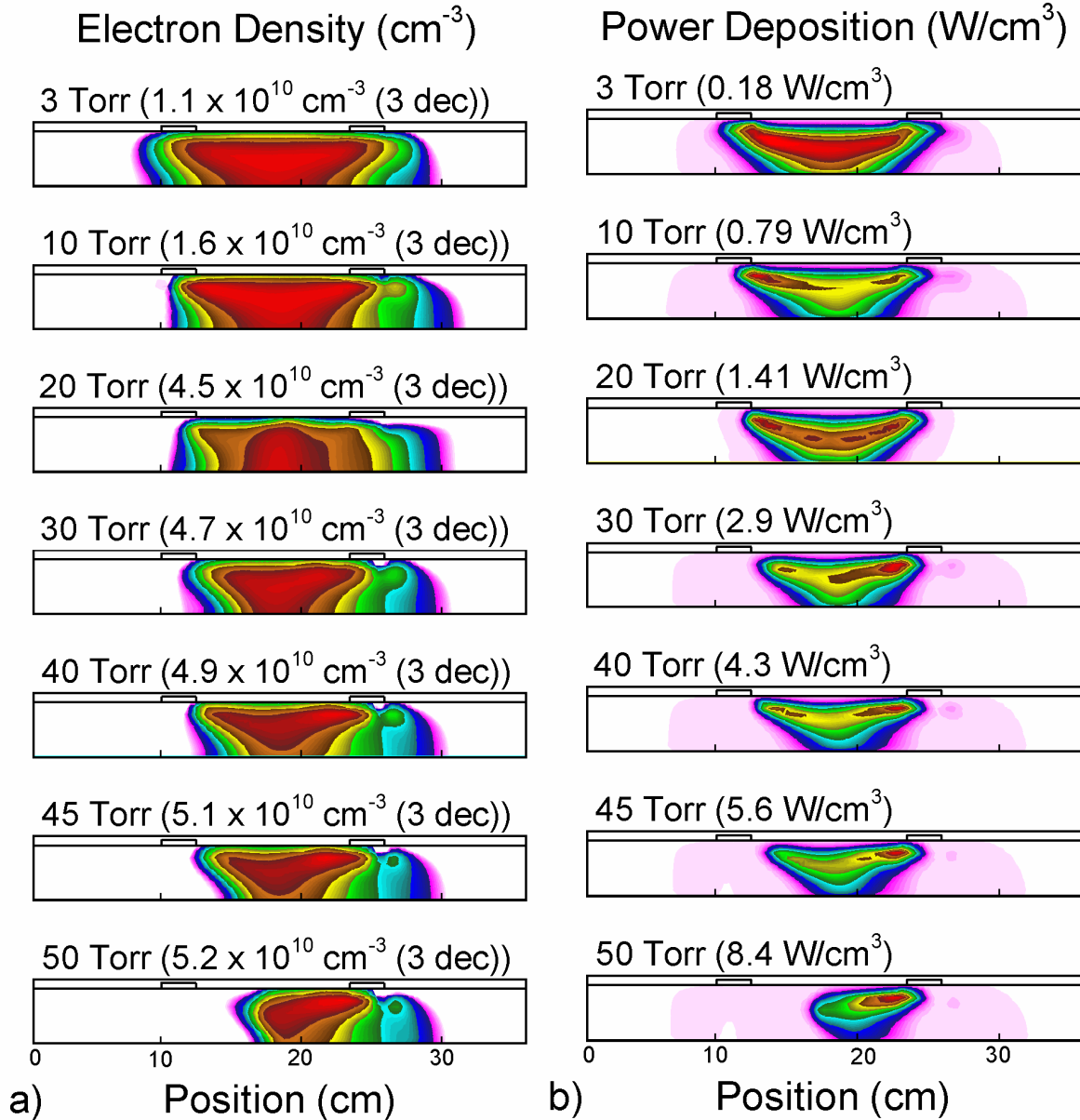


MIN  MAX

- Change in enthalpy from 3-body reactions increases with pressure.
- Heating accumulates in moving downstream.
- Rarefaction increases with pressure
- Localized rarefaction near downstream electrode where power is maximum.
- 10 to 50 Torr, 133 to 670 W, 25 MHz, 0.3 eV/molecule

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DIFFUSIVE TO CONSTRICTED DISCHARGE

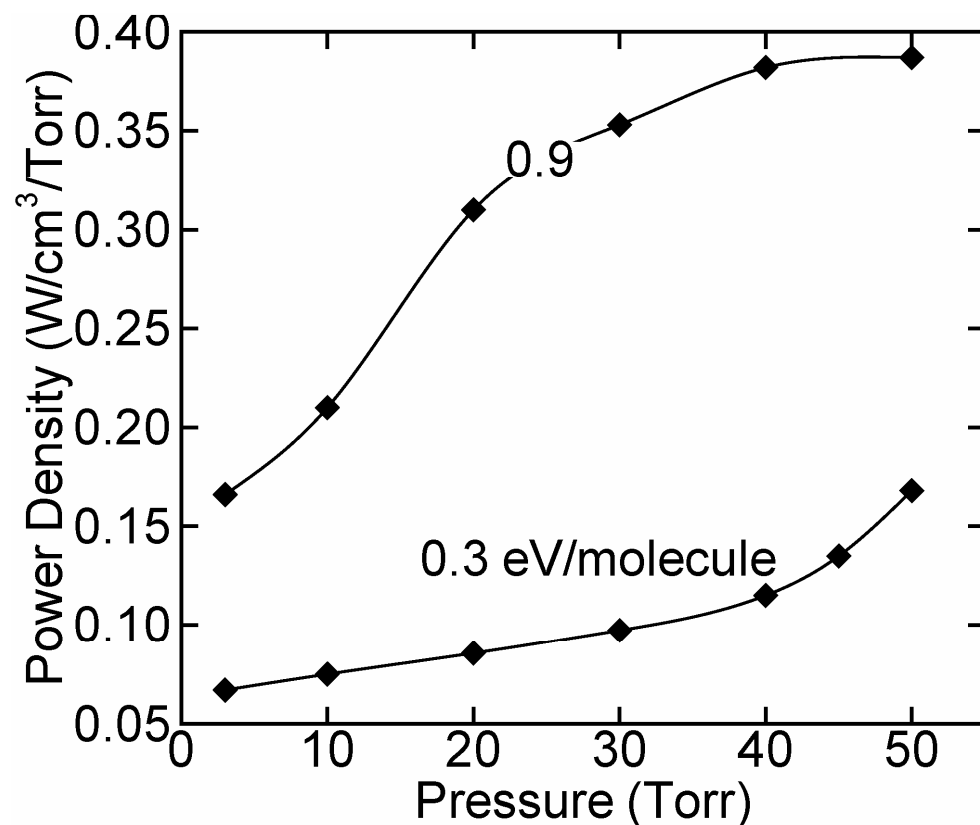


- Power deposition and electron density gradually constrict due to rarefaction of the gas.

- 10 to 50 Torr, 133 to 670 W, 25 MHz, 0.3 eV/molecule

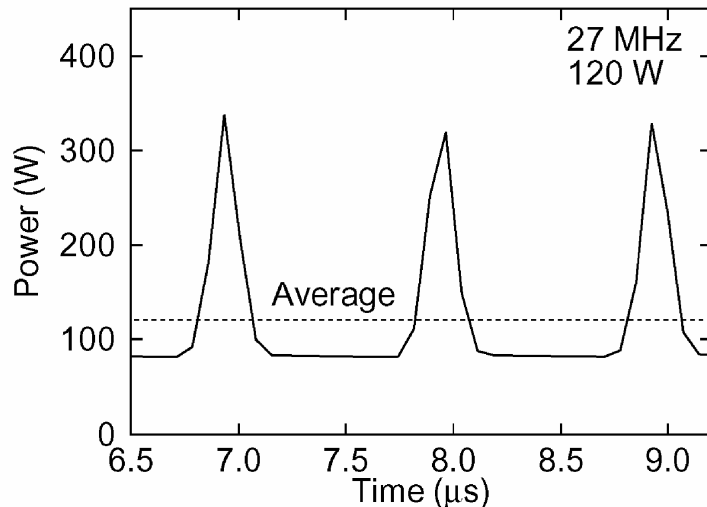
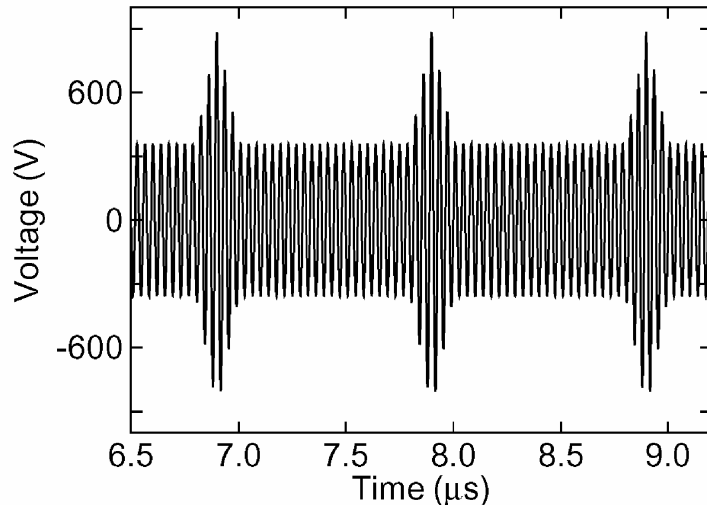
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POWER DEPOSITION



- **Rapid increase in pressure normalized power deposition indicates a discharge constriction.**
- **For high energy loadings constriction occurs at lower pressure.**

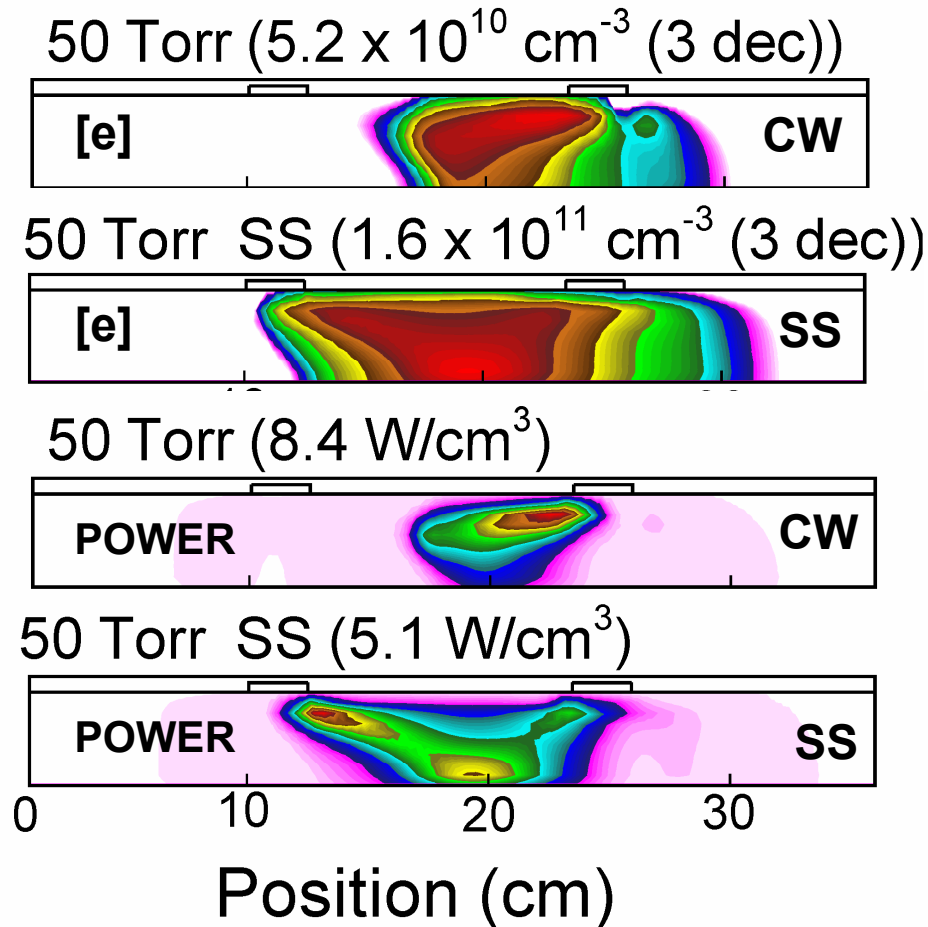
NON-SELF SUSTAINED DISCHARGES: SPIKER SUSTAINER



- Short high power (spiker) pulse is followed by plateau of lower power (sustainer). Excess ionization in “afterglow”.
- Spiker sustainer (SS) consists of modulated rf excitation.
- 27 MHz, He/O₂ = 70/30, 3 Torr

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IMPROVEMENT WITH SPIKER-SUSTAINER (SS)

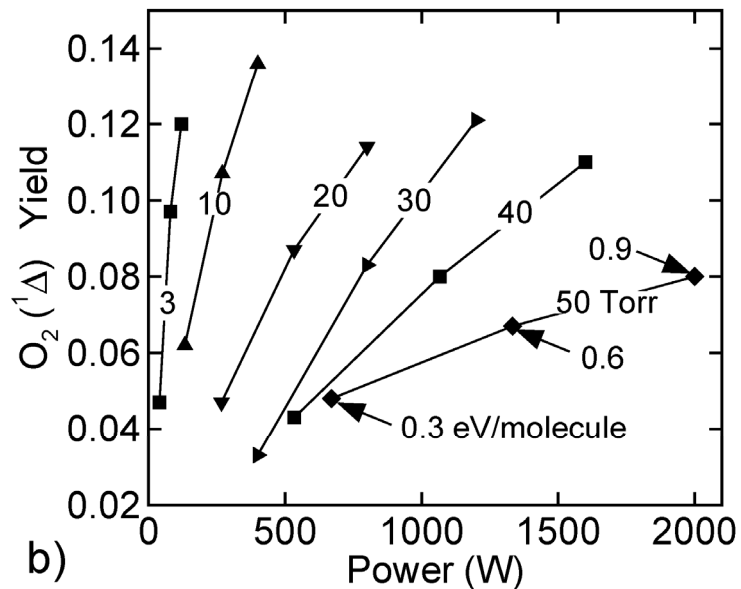
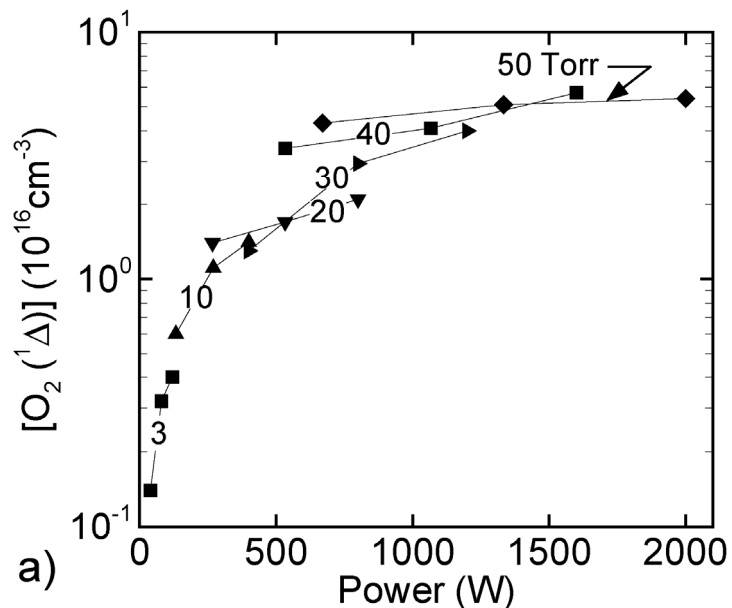


MIN  MAX

- CW discharge becomes inhomogeneous at high pressures and high energy loading.
- In SS discharges, ionization is produced by a high-voltage pulse, which is turned off before ionization instabilities develop.
- SS discharge remains stable to higher pressures and power inputs.

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O₂(¹Δ) and Yield



SUMMARY OF PRESSURE SCALING

- Absolute O₂(¹Δ) production increases with power for any given pressure.
- Yield increases with power for any given pressure.
- Yield at a given power generally decreases with pressure.
- He/O₂=70/30, 6 lpm, 25 MHz, 3-50 Torr, 0.3, 0.6, 0.9 eV/molecule

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CONCLUDING REMARKS

- **Oxygen-iodine kinetics in flowing afterglows for eCOIL has been computationally investigated.**
- **High pressure operation should produce larger densities of $O_2(^1\Delta)$ but densities and yields usually scale sub-linearly with pressure (or decrease).**
- **Careful management is required:**
 - **O_3 and atomic oxygen densities: Left unchecked, quenching at high pressure is a rate limiting step**
 - **Influence of O-atom quenching can be minimized by NO injection.**
 - **Gas temperature: Active cooling increases O_3 formation**
 - **Stability: Asymmetries in excitation due to rarefaction may trigger constrictions.**